

# Next Generation Lipid-Based Excipients: Solid State - Stability Relationship

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## Background and Objectivie

Lipid-based excipients (LBE) are known as naturally occurring materials, being easily available, biocompatible, and attractive for manufacturing of a broad range of dosage forms. The main challenge is their solid state transformation, thus instability of their final product.

The purpose of this work was:

1. to design an advanced group of LBEs; polyglycerol esters of fatty acids (PGFAs) (Witepsol PMF®) (Figure 1), with stable solid state to overcome the stability issues of conventional LBEs,
2. structure-function analysis of selected molecules for their application as hot melt coating (HMC) excipients, matrix agent for direct compaction (DC) of extended release tablets and matrix agent for manufacturing of solid lipid nanosuspensions (SLN).

## Lipid-Based Excipients

PGFAs are oligomeric hydroxyethers of glycerol, fully or partially esterified with fatty acids (Figure 1). Tuning the number of glycerol moieties, fatty acids chain length and free hydroxyl groups per molecule results in different HLB values, melting points, melt viscosities (1).

## Application of PGFAs in Pharmaceutical Production; Case Studies

The DSC, SAXS and WAXS data revealed the crystallization of all selected lipids into a stable monophasic system in  $\alpha$ -form without transformation and crystal growth during storage at 40 °C (1).

Dosage form	PGm	Cn	Esterification	Witepsol® PMF	HLB	Melting Point (°C)
SLN	PG2	C18	Full	282	2.6	59.4
Matrix tablets (DC)	PG3	C22	Partial	123	3.7	74.4
	PG2	C22	Full	222	1.8	72.5
Hot melt coated particles	PG3	C16/C18	Partial	1683	5.1	54.2
	PG6	C18	Partial	186	6.2	59.26
	PG4	C18	Partial	184	5.6	60.33

PG4C18 Partial  
WITEPSOL® PMF 184

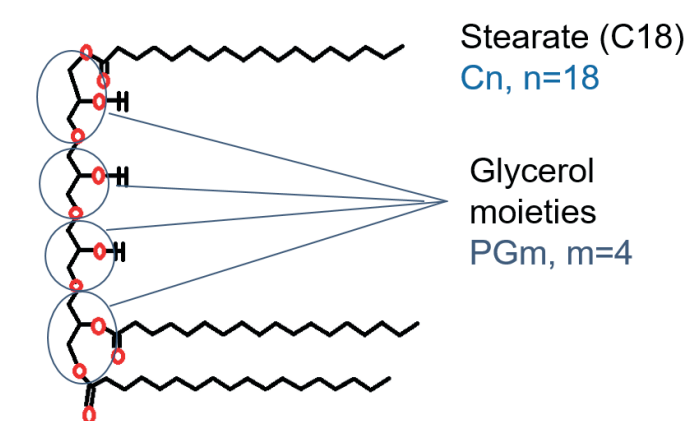


Figure 1- Chemical structure of selected PGFAs for using as LBE to manufacture matrix tablets, hot melt coated particles and SLNs, depiction of the chemical structure of PG4-C18 partial

## Hot Melt Coating (2)

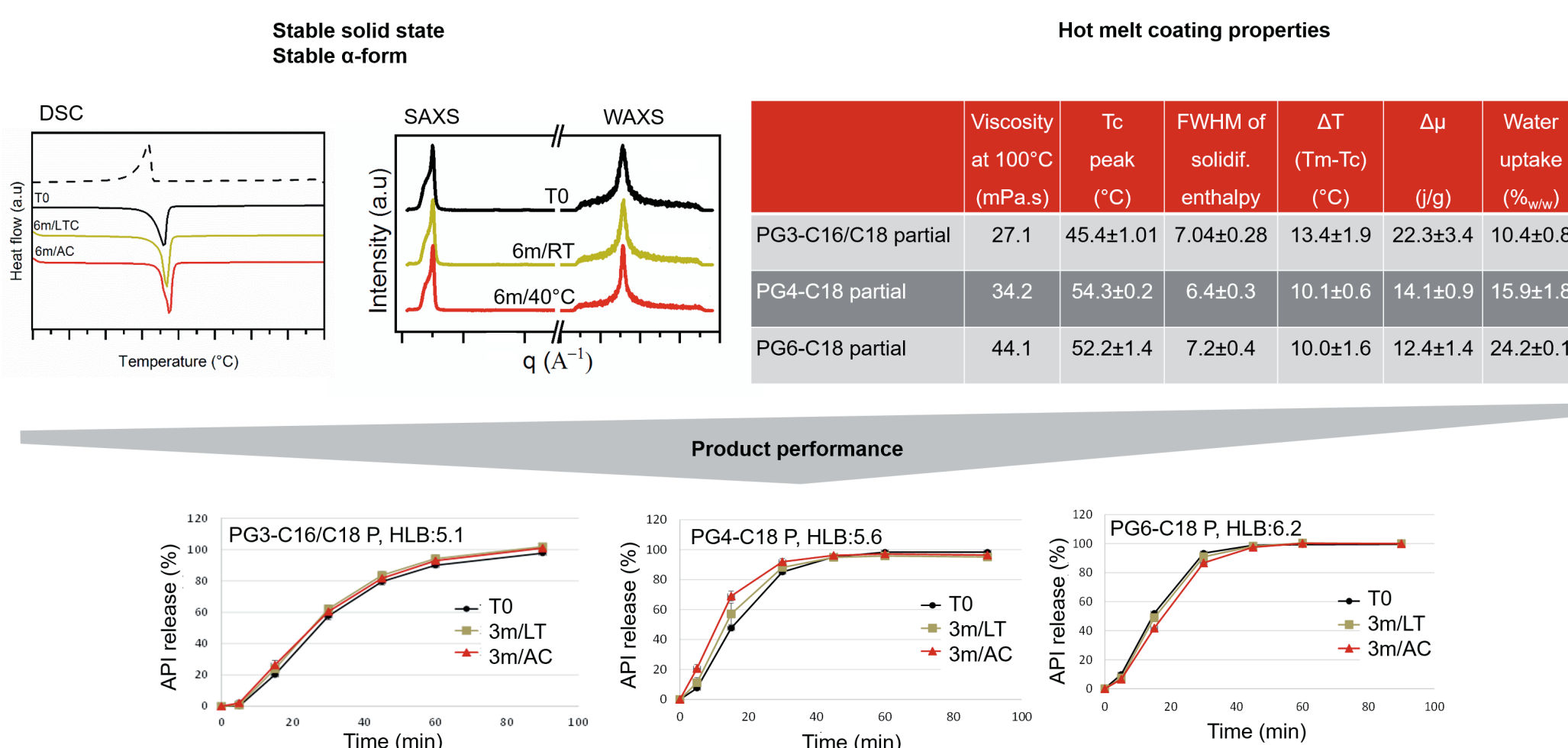


Figure 2- Solid state and hot melt coating properties of selected PGFAs, release profile of N-acetylcysteine from coating after manufacturing and after 3 months storage. AC= accelerated condition, LT = long term, RT = room temperature, Tc = crystallization temperature, Tm = melting temperature,  $\Delta\mu$  = driving force for nucleation

## Direct compaction (3)

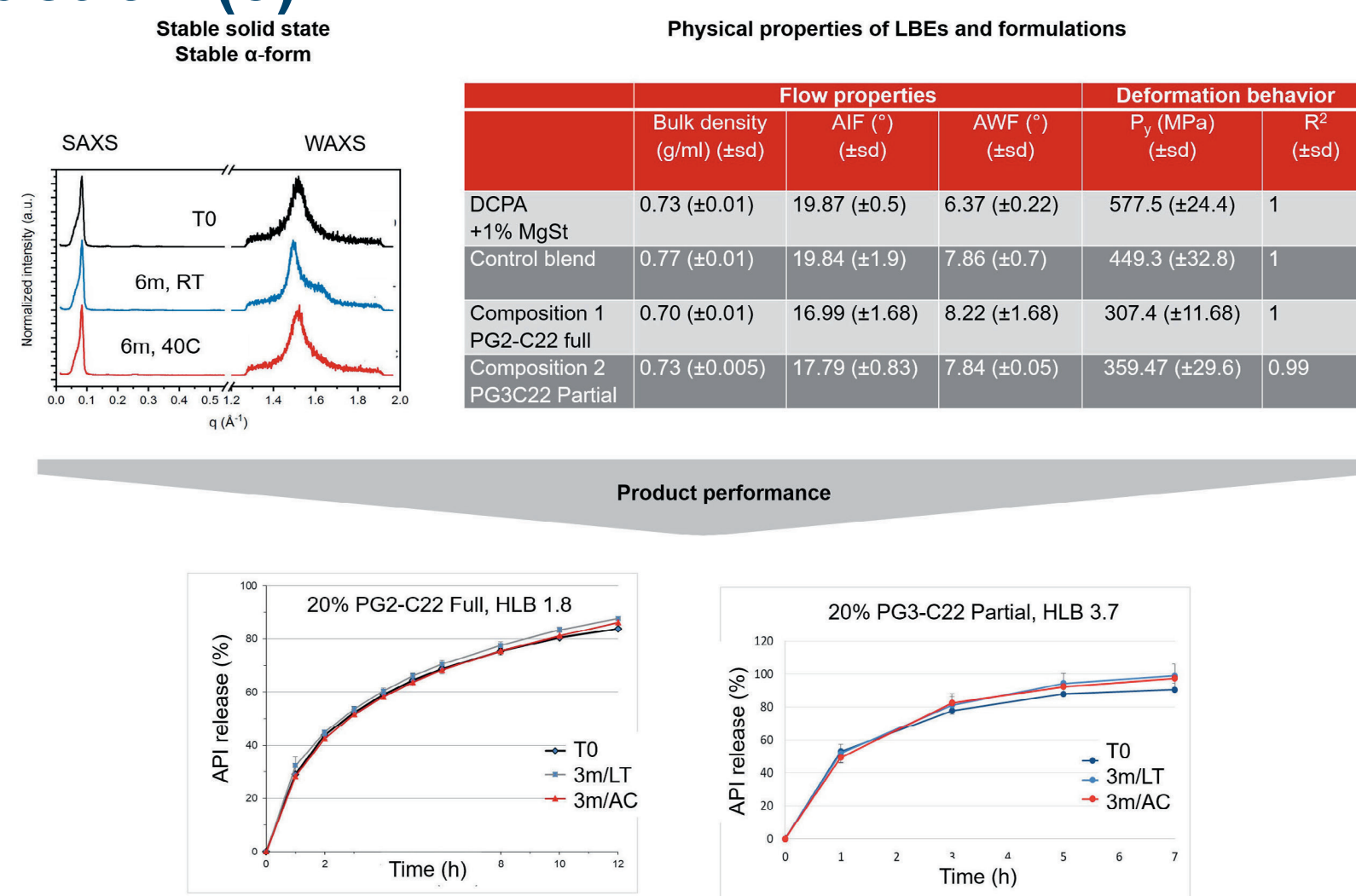


Figure 3- Solid state and tableting properties of selected PGFAs, release profile of metformin HCl from matrix tablets after manufacturing and after 3 months storage. AIF = angle of internal friction, AWF = angle of wall friction, Py = yield pressure

## Solid lipid nanosuspension (4)

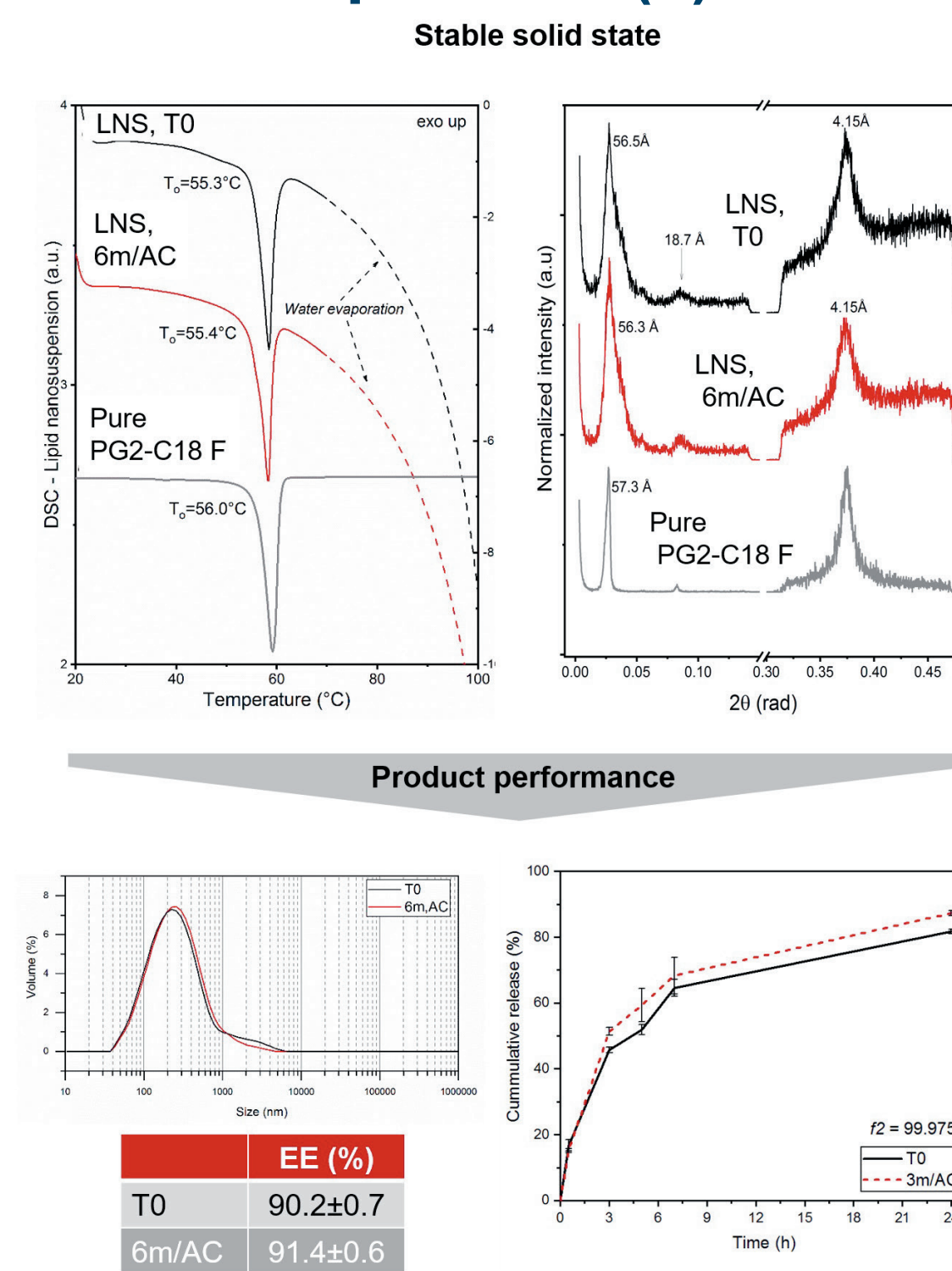


Figure 4- Solid state and thermal behavior of pure PG2-C18 F and the SLNs after preparation and after 6 months storage under accelerated conditions (25°C). PSD, release profile of dexamethasone from LSN and entrapment efficiency after preparation and 6 months storage.

## Conclusion and Outlook

The monophasic crystallization of PGFAs into stable  $\alpha$ -form and no crystallite growth lead to stable performance of developed product. The release profile can be tailored based on the HLB of selected PGFAs. Data on further applications for spray-dried powder for inhalation and extrusion are available.

## References

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